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Chapter 18 - Nuclear Chemistry 289 Key Ideas Answers 14. Because protons and neutrons reside in the nucleus of atoms, they are called nucleons. 16. There are two forces among the particles within the nucleus. The first, called the electrostatic force, is the force between electrically charged particles. The second force,

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Answer. The thermal decomposition of calcium carbonate, with the desired product being calcium oxide and waste product

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being carbon dioxide is shown like this: $\text{CaCO}_3 (\text{s}) \rightarrow \text{CaO} (\text{s}) + \text{CO}_2 (\text{g})$ The total mass of all the reactants = mass of CaCO_3 = 100 g. The mass of desired product = mass of CaO = 56 g.

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HClO_4 is the stronger acid because, in a series of oxyacids with similar formulas, the higher the electronegativity of the central atom, the stronger is the attraction of the central atom for the electrons of the oxygen(s). The stronger attraction of the oxygen electron results in a stronger attraction of oxygen for the electrons in the O-H bond, making the hydrogen more easily released.

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Chapter 18: Organic Chemistry. organic chemistry. carbon. 4. valence electrons. the study of carbon containing compounds. is in more compounds than any other element. carbon always forms ___ strong covalent bonds. carbon is composed of 4 _____.

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Chemistry Ch-2 (Question-answer) 1) Fill in the blanks. a) Non-metals are generally soft but diamond is an exception. b) Among non-metals, nitrogen does not react with bases at all. c) Hydrogen is the only non-metal included in the reactivity series of metals. d) Sulphur is also used in hardening of rubber. e) The impurities that are added to semiconductors to alter their electrical ...

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ox: $[Ag \rightarrow Ag^+ + e^-] \times 2$; red: $Cu^{2+} + 2e^- \rightarrow Cu$. balanced: $2Ag(s) + Cu^{2+}(aq) \rightarrow 2Ag^+(aq) + Cu(s)$ $E^\circ = E^\circ_{ox} + E^\circ_{red} = -0.80V + 0.34V = -0.46V$. Not spontaneous. ox: $[Ni(s) \rightarrow Ni^{2+} + 2e^-] \times 5$; red: $[5e^- + 8H^+ + MnO_4^- \rightarrow Mn^{2+} + 4H_2O] \times 2$.

Chapter 18 Nuclear Chemistry

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